

Name:

Class:

Year 6 Signature Work Inquiry:

Cool Water

Central inquiry journal



SUSTAINABLE DEVELOPMENT GOALS

3 GOOD HEALTH AND WELL-BEING



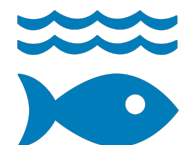
6 CLEAN WATER AND SANITATION



12 RESPONSIBLE CONSUMPTION AND PRODUCTION



14 LIFE BELOW WATER





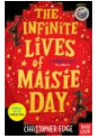
Year 6 Signature Work Heroic Inquiry Journey



Author visit:
Christopher Edge



Reading science-rich fiction:
The Infinite Lives of Maisie Day
by Christopher Edge



Reading scientific non-fiction about
heat transfer



Working in teams to plan and
deliver a persuasive campaign

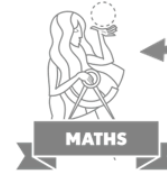
Reflective writing: CREST application
Creating a poster to summarise and
reflect on the year's work



Practical
investigation into
insulating materials



Designing and
making insulating
covers for reusable
water bottles



Plotting and
interpreting graphs
of the results from
the Science
experiments



Collaborating in groups using Office 365

Using Word and PowerPoint to create
effective presentations and posters



CREST Bronze
Award
submission

In Art/DT this term you will be making an insulating cover for your reusable water bottle.

How does this link to these SDGs?

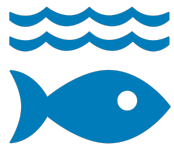
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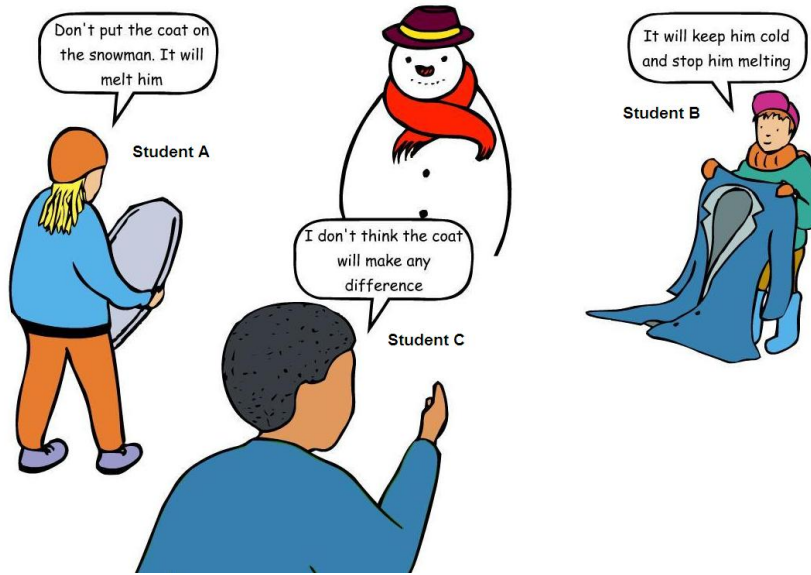


12 RESPONSIBLE CONSUMPTION AND PRODUCTION



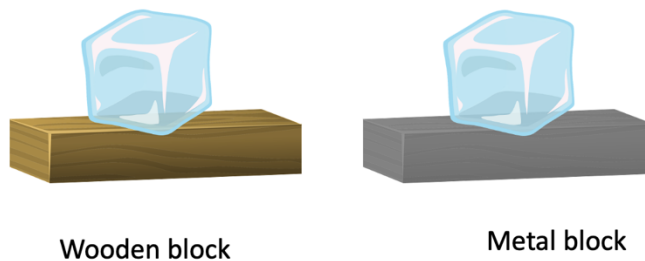
Year 6 Signature Work Inquiry: Cool water bottles

The Snowman's Coat



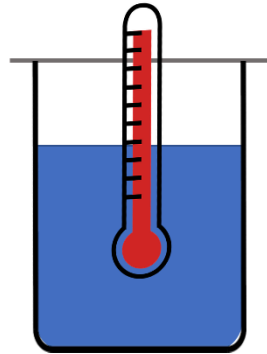
	The coat will make him melt faster	The coat will make him melt slower	The coat will make no difference
My first thought is...			
After I've spoken to my partner I think...			
After our experiment I think...			

Which will melt faster?

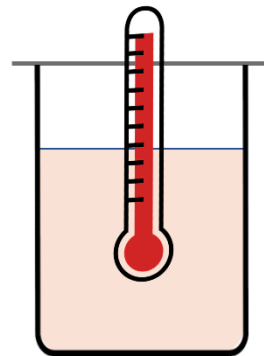


	The ice on the wooden block will melt faster	The ice on the metal block will melt faster	They will both melt at the same rate
My first thought is...			
After I've spoken to my partner I think...			
After our experiment I think...			

Which will change in temperature the most?



Ice water



Hot water

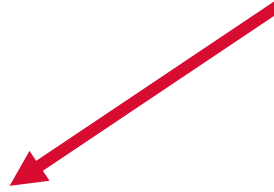
	Starting temperature (°C)	Temperature after 10 minutes (°C)	Change in temperature (°C)
Ice water			
Hot water			

Which one changed temperature the most in 10 minutes?

If we are trying to find out which materials are good insulators, does it matter if we use hot or cold water (as long as we always do the same thing?)

Which do you think we should use in our experiments? Why?

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1. What is heat?

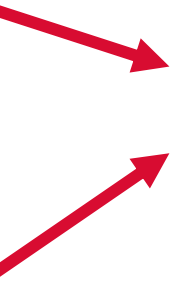
2. How do the atoms and molecules in a hot object move (compared to a colder object)?

3. THINK! if a hot material (e.g. a warm plate) is touching a cold material (e.g. an ice cube) how do you think heat energy will be passed from one to the other?

4. THINK: Given your answer to question 2, why do you think hot materials expand (take up more space) compared to cold ones?

6. What is temperature?

7. Explain in your own words how a thermometer works?



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1. Name the three processes by which heat is transferred

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2. CONVECTION: Convection ONLY happens in fluids. What are fluids?

3. CONVECTION: What happens to the atoms and molecules in a fluid when it gets hotter?

4. CONVECTION: Do hot fluids float or sink? What about cold fluids?

5. THINK: Look at question 1. The walls of your water bottle/ measuring cylinder are solid. Do you think heat energy can travel through them **by convection**? Why/why not?

6. CONDUCTION: Explain in your own words how heat energy travels through solids.

7. CONDUCTION: Look at the box about INSULATION. Is trapped air a good or bad conductor of heat energy?


How do clothes keep us warm?

8. Trapped air is a really good INSULATOR. Think of some materials to wrap your water bottle in that might contain pockets of trapped air.


10. RADIATION: What is the name of the special kind of radiation that carries heat energy?

11. RADIATION: What kinds of objects give out heat radiation?

12. RADIATION: Can radiation travel through places where there are no atoms (like space)?



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SOURCE B: Dorling Kindersley (2018) *How Science Works*. p.118-9

Heat transfer notes

Read the articles and make some notes in each red box. Don't forget to say which source you have got the information from (e.g. Source A). Then **in the water bottle outline** suggest how you can use what you have learnt to keep the contents of your bottle cool (or hot).

Heat and temperature

Conduction (and insulation)

Convection (how about a diagram?)

Radiation

Tick the sources you used for your information heat transfer

- SOURCE A:** Claybourne, A. (2013) *My First Book of Science*. P.44-5
- SOURCE B:** Dorling Kindersley (2018) *How Science Works*. p.118-9
- SOURCE C:** Khan, S. & Gillespie, L.J. (2012) *Usborne Junior Illustrated Science Dictionary*. P.94-
- SOURCE D:** Parsons, J. (ed) (2004) *DK Illustrated Family Encyclopedia (G)*. p.416-7
- SOURCE E:** Hart-Davies, A. (ed) (2021) *Science: The Definitive Visual Guide*. p.471
- SOURCE F:** Dorling Kindersley (2020) *The Visual Encyclopaedia*. p.185

Conclusions

Most of the heat energy will travel out of my bottle by (*conduction, convection or radiation?*)

So I need to slow this down by wrapping my bottle in a really good...

Many of good insulators lots of...

So I think that materials like these might be good at keeping my water cool:

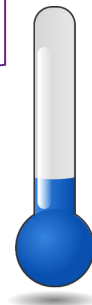
Science, Maths & ICT: Reflection

What **problems** did I (or the group) have with the Science experiments?
(4.4)

What decisions did we make during our experiments? (4.2)

How did I **overcome** these problems? (4.4)

How did this affect the way our experiments turned out? (3.2)



What did I learn? If I had to do these experiments again, how would I do them better next time? (3.3)

Date

Art & DT: Reflection

What were the hardest things about making my bottle cover? (4.4)

What decisions did I make about my bottle cover? (4.2)



How did I **overcome** these problems? (4.4)

How did this affect the way my project turned out? (3.2)

What did I learn? If I had to do a sewing project like this again, how would I do it better next time? (3.3)